



Designation: D 6556 – 00a

Standard Test Method for Carbon Black—Total and External Surface Area by Nitrogen Adsorption¹

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1. Scope

1.1 This test method covers the determination of the total surface area by the Brunauer, Emmett, and Teller (B.E.T. NSA) theory of multilayer gas adsorption behavior using multipoint determinations and the external surface area based on the statistical thickness surface area method.

1.2 The values stated in SI units are to be regarded as the standard. The values given in parentheses are for information only.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* (The minimum safety equipment should include protective gloves, sturdy eye and face protection).

2. Referenced Documents

2.1 ASTM Standards:

- D 1799 Practice for Carbon Black—Sampling Packaged Shipments²
- D 1900 Practice for Carbon Black—Sampling Bulk Shipments²
- D 3324 Practice for Carbon Black—Improving Test Reproducibility Using ASTM Reference Blacks²
- D 4483 Practice for Determining Precision for Test Method Standards in the Rubber and Carbon Black Industries²

3. Summary of Test Method

3.1 The total and external surface areas are measured by evaluating the amount of nitrogen adsorbed, at liquid nitrogen temperature, by a carbon black at several partial pressures of nitrogen. The adsorption data is used to calculate the NSA and STSA values.

4. Significance and Use

4.1 This test method is used to measure the total and external surface area of carbon blacks based on multipoint

nitrogen adsorption. The NSA measurement is based on the B.E.T. theory and it includes the total surface area, inclusive of micropores, pore diameters less than 20 Å. The external surface area, based on the statistical thickness method (STSA), is defined as the specific surface area that is accessible to rubber.

5. Apparatus

5.1 *Multipoint Static-Volumetric Gas Adsorption Apparatus*, with Dewar flasks and all other accessories required for operation.

5.2 *Sample Cells* that, when attached to the adsorption apparatus, will maintain isolation of the sample from the atmosphere equivalent to a helium leak rate of $<10^{-5}$ cm³/min, per atmosphere of pressure difference.

5.3 *Balance, Analytical*, with 0.1 mg sensitivity.

5.4 *Heating Mantle or Equivalent*, capable of maintaining a temperature of $300 \pm 10^\circ\text{C}$.

5.5 *Oven, Gravity Convection*, capable of maintaining a temperature of $125 \pm 10^\circ\text{C}$.

6. Reagents

6.1 *Liquid nitrogen*, 98 % or higher purity.

6.2 *Ultra-high purity nitrogen gas*, cylinder or other source of prepurified nitrogen gas.

6.3 *Ultra-high purity helium gas*, cylinder or other source of prepurified helium gas.

7. Sampling

7.1 Samples may be taken in accordance with Practice D 1799 and Practice D 1900.

8. Sample Preparation Procedure

8.1 Dry a portion of carbon black at 125°C for 1 h. If the carbon black is known to be substantially free of moisture, or subsequent preparation steps are known to be adequate for moisture removal, then this step may be omitted.

8.2 Condition an empty sample cell for a minimum of 10 minutes at the same conditions intended for degassing the sample. Weigh the empty sample cell to the nearest 0.1 mg and record the mass.

8.3 Weigh approximately 0.4 g of the carbon black into the sample cell.

NOTE 1—For carbon black powder samples, add enough carbon black

¹ This test method is under the jurisdiction of ASTM Committee D24 on Carbon Black and is the direct responsibility of Subcommittee D24.21 on Adsorptive Properties of Carbon Black.

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² *Annual Book of ASTM Standards*, Vol 09.01.

to give a depth of approximately 2 in. in straight wall sample tubes, or approximately 0.4 g for bulb-type sample cells.

8.4 Flow Degassing:

8.4.1 Open the gas control valve and insert the delivery tube into the sample tube, and allow purging with either helium or nitrogen for a minimum of 1 minute.

8.4.2 Place a heating mantle or other source of heat around the sample cell and degas the sample at $300 \pm 10^\circ\text{C}$ for $\frac{1}{2}$ h or longer to ensure that all traces of moisture condensing in the top of the tube are absent. The minimum degassing time that gives a stable surface area (that is, a surface area that does not increase with additional degassing) may be used for degassing.

8.4.3 Once the typical degassing times have been determined, future samples can be degassed on the basis of time alone, if desired, allowing a reasonable margin of excess time. Some samples will be found to require less than $\frac{1}{2}$ h, especially if moisture exposure has been minimal. In these cases, the minimum time that gives a stable surface area may be used for degassing.

8.4.4 After degassing, the sample tube may be moved directly to the analyzer. Otherwise, remove the sample tube from the heat source and continue the flow of purging gas until it is ready for analysis.

8.4.5 Go directly to Section 9 and continue the remaining steps of the procedure.

8.5 Vacuum Degassing:

8.5.1 With the apparatus at atmospheric pressure, place the sample cell containing the carbon black onto the degassing apparatus.

8.5.2 Begin the degassing procedure as appropriate for the apparatus.

8.5.3 Place a heating mantle or other source of heat around the sample cell and degas the sample at $300 \pm 10^\circ\text{C}$ for $\frac{1}{2}$ h or longer as required to obtain and hold a pressure less than 1.4 Pa (10 μm Hg).

8.5.4 Once the typical degassing times have been determined, future samples can be degassed on the basis of time alone, if desired, allowing a reasonable margin of excess time. Some samples will be found to require less than $\frac{1}{2}$ h, especially if moisture exposure has been minimal. In these cases, the minimum time that gives a stable surface area may be used for degassing.

8.5.5 Go directly to Section 9 and continue the remaining steps of the procedure.

9. Measurement Procedure

9.1 Refer to the user's manual or specific instructions for the multipoint gas adsorption analyzer to be used, and become thoroughly familiar with the procedures.

9.2 Fill the Dewar with liquid nitrogen and allow it to reach temperature equilibrium, preferably 0.5 to 1 h.

9.3 Accurately determine the saturation pressure of the liquid nitrogen bath by running replicate determinations until two consecutive saturation pressure values agree within 0.13 kPa (1 mm Hg).

9.4 Determine the free space of the sample cell by measurement with helium or by calculation using an assumed carbon black density of 1.9 g/cm³.

9.5 Obtain a minimum of five data points evenly spaced in

the 0.1 to 0.5 relative pressure (P/P_o) range. For some tread carbon blacks, particularly N100 and N200 series, it is necessary to measure two additional data points, 0.05 and 0.075, in order to increase the accuracy of the NSA measurement. A data point consists of the relative pressure of equilibrium and the total amount of nitrogen gas adsorbed by the sample at that relative pressure.

9.6 Determine the mass of the cell with dry sample to the nearest 0.1 mg. This may be done before or after measuring nitrogen adsorption. Avoid inconsistent use of helium, as a buoyancy error of 1 mg/cm³ of cell volume can occur. As an alternative, the carbon black mass may be determined directly by pouring it from the sample cell into a tared weighing pan, taking care to remove all of the carbon black.

10. Calculation

10.1 Most automated instruments will perform the following computations at the completion of the analysis. The user must verify that the internal computations conform to the following method.

10.2 Sample Mass:

$$\text{sample mass (dried)} = (\text{mass of cell} + \text{sample}) - (\text{mass of cell}) \quad (1)$$

Record masses to nearest 0.1 mg.

10.3 Volume of Nitrogen Adsorbed

10.3.1 Calculate total volume of nitrogen adsorbed per gram of sample to the nearest 0.0001 cm³/g as follows:

$$V_a = \frac{\text{Volume of Nitrogen for each dosing in cm}^3}{\text{sample mass in g}} \quad (2)$$

10.4 Nitrogen Surface Area:

10.4.1 Determine the nitrogen surface area (NSA) using a B.E.T. plot from the Brunauer, Emmett, and Teller³ equation as follows:

$$\frac{P}{V_a(P_o - P)} = \frac{1}{V_m C} + \frac{C - 1}{V_m C} \times \frac{P}{P_o} \quad (3)$$

where:

P = manometer pressure in kPa,

P_o = saturation vapor pressure of nitrogen in kPa,

V_m = volume of nitrogen per gram that covers one monomolecular layer in standard cm³/g, and

C = a constant that is a function of average heat of adsorption of the monomolecular layer,

10.4.2 Plot P/P_o on the X-axis versus $\frac{P}{V_a(P_o - P)}$ on the Y-axis, for data sets having P/P_o in the range of 0.05 to 0.30 (linear region of B.E.T. equation).

10.4.3 The data points (three or more) that give the best straight line are used to calculate the slope and y-intercept. The slope and y-intercept are used to calculate the surface area. For examples of how to select the proper relative pressure range, see Table 1.

10.4.4 As an alternative, the interpretation of the proper relative pressure can generally be simplified by specifying the following pressure ranges for the various carbon black types:

³ Brunauer, Emmett and Teller, Journal of the American Chemical Society, Volume 60, 1938, p. 309.

TABLE 1 Example of NSA Data Analysis

N121 ^A				
Raw Data		Calculation		
P/P _o	Vol. Ads., cm ³ /g	Rel. Press. Range	Correlation Coefficient	NSA, m ² /g
0.0500	26.716
0.1000	29.753
0.1500	32.313	0.05–0.15	0.999981	123.9
0.2000	34.692	0.05–0.20	0.999992	124.0
0.2500	37.110	0.05–0.25	0.999990	123.6
0.3000	39.641	0.05–0.30	0.999935	122.8
N326 ^B				
Raw Data		Calculation		
P/P _o	Vol. Ads., cm ³ /g	Rel. Press. Range	Correlation Coefficient	NSA, m ² /g
0.0500	16.675
0.1000	18.318
0.1500	19.859	0.05–0.15	0.999960	75.6
0.2000	21.426	0.05–0.20	0.999948	76.3
0.2500	23.035	0.05–0.25	0.999964	76.6
0.3000	24.751	0.05–0.30	0.999979	76.6
N683 ^B				
Raw Data		Calculation		
P/P _o	Vol. Ads., cm ³ /g	Rel. Press. Range	Correlation Coefficient	NSA, m ² /g
0.0500	8.194
0.1000	9.113
0.1500	9.945	0.05–0.15	0.999939	38.2
0.2000	10.739	0.05–0.20	0.999950	38.5
0.2500	11.543	0.05–0.25	0.999972	38.6
0.3000	12.364	0.05–0.30	0.999973	38.4

^AThe most accurate NSA is measured between 0.05 and 0.20 relative pressure.

^BThe most accurate NSA is measured between 0.05 and 0.30 relative pressure.

	BET Range
N300 and Carcass Grades	0.1–0.3
N100 and N200	0.05–0.2
Carbon Blacks > 130 m ² /g	0.05–0.1

It is the responsibility of the operator to assure that these guidelines are appropriate for their samples.

10.4.5 A B.E.T. plot that yields a negative y-intercept could be indicative of the presence of micropores (<2 nm diameter), but other factors can produce a negative y-intercept. The surface area is calculated from three or more points within the pressure range that yields the highest correlation coefficient and a positive y-intercept.

10.4.6 Calculate the nitrogen surface area to the nearest 0.1 m²/g as follows:

$$\text{Surface area (m}^2\text{/g)} = V_m \times 4.35 \quad (4)$$

where:

$$V_m = \frac{1}{B + M}$$

$$B = \text{Y-axis intercept, } \pm 10^{-5}$$

$$M = \text{slope of the straight line, } \pm 10^{-5}, \text{ and}$$

$$4.35 = \frac{\text{area occupied by 1 cm}^3 \text{ of nitrogen} = (6.02 \times 10^{23})(16.2 \times 10^{-20})}{22400}$$

$$6.02 \times 10^{23} = \text{Avogadro's number,}$$

$$16.2 \times 10^{-20} = \text{area of nitrogen molecule in m}^2, \text{ and}$$

22400 = number of cm³ occupied by one mole of gas at STP.

10.5 Statistical Thickness Surface Area:

10.5.1 Determine the STSA⁴ of the black using a plot of the volume of nitrogen gas adsorbed per gram of sample at STP (V_a) versus the statistical layer thickness (t).

10.5.2 Prepare the V_a - t plot by plotting t (Å) on the X axis versus V_a (cm³/g at STP) on the Y axis, for data sets having P/P_o equally spaced in the range of 0.2 to 0.5.

where:

$$t = \text{statistical layer thickness of carbon black} = 0.88(P/P_o)^2 + 6.45(P/P_o) + 2.98$$

10.5.3 Determine the slope of the V_a - t plot using standard linear regression.

10.5.4 Calculate the STSA to the nearest 0.1 m²/g as follows:

$$STSA = M \times 15.47 \quad (5)$$

where:

M = slope of the V_a - t plot, and
15.47 = a constant for the conversion of nitrogen gas to liquid volume, and conversion of units to m²/g.

10.5.5 In instances where the V_a - t plot yields a negative intercept, report the STSA value as being equivalent to the NSA value. This prevents STSA values from being higher than the NSA values, as this is a theoretical impossibility.

11. Report

11.1 Report the following information:

- 11.1.1 Proper sample identification,
- 11.1.2 Number of data points and relative pressures used to obtain both NSA and STSA,
- 11.1.3 The sample mass to the nearest 0.1 mg, and
- 11.1.4 The NSA and/or STSA of the sample reported to the nearest 0.1 m²/g.

12. Precision and Bias

12.1 This precision and bias has been prepared in accordance with Practice D 4483. Refer to this practice for terminology and other statistical details.

12.2 The precision results in this precision and bias give an estimate of the precision as described below. The precision parameters should not be used for acceptance or rejection testing of any group of materials without documentation that they are applicable to those particular materials and the specific testing protocols that include this test method.

12.3 Nitrogen Surface Area (NSA)

12.3.1 A Type 1 inter-laboratory precision program was conducted in 1991. Both repeatability and reproducibility represent short-term testing conditions. Seven laboratories tested six carbon blacks (SRB A-4, B-4, C-4, D-4, E-4, and F-4) twice on each of two different days. Therefore, $p = 7$, $q = 6$, and $n = 4$. A test result is the value obtained from a single determination. Acceptable difference values were not measured.

⁴ Magee, R. W., Rubber Chemistry and Technology, Volume 68, No. 4, 1995, p. 590.

NOTE 2—The values for the SRB-5 series blacks are listed in Table 2.

12.3.2 The results of the precision calculations are given in Table 3 with the material arranged in ascending “mean level” order.

12.3.3 The precision for the pooled values for NSA may be expressed as follows:

12.3.4 *Repeatability*—The repeatability, (*r*), of the NSA Method has been established as 1.92 %. Two single test results (or determinations) that differ by more than 1.92 % must be considered suspect and dictate that some appropriate investigative action be taken.

12.3.5 *Reproducibility*—The reproducibility, (*R*), of the NSA Method has been established as 4.38 %. Two single test results (or determinations) produced in separate laboratories that differ by more than 4.38 % must be considered suspect and dictate that some appropriate investigative or technical action be taken.

12.4 Statistical Thickness Surface Area (STSA)

12.4.1 A Type 1 interlaboratory precision program was conducted in 1995. Both repeatability and reproducibility represent short-term testing conditions. Twenty-one laboratories tested six carbon blacks (SRBs A-5 through F-5) twice on each of two different days. A test result is the value obtained from a single determination. Acceptable difference values were not measured (see Table 4).

12.4.2 The results of the precision calculations are given in Table 4 with the material arranged in ascending “mean level” order.

12.4.3 The precision for the pooled values for STSA may be expressed as follows:

12.4.4 *Repeatability*—The pooled repeatability of the STSA Method has been established as $2.06 \times 10^3 \text{ m}^2/\text{kg}$ ($2.06 \text{ m}^2/\text{g}$). Two single test results (or determinations) that differ by more than $2.06 \times 10^3 \text{ m}^2/\text{kg}$ ($2.06 \text{ m}^2/\text{g}$) must be considered suspect, that is, to have come from different sample populations. Such a decision dictates that some appropriate action be taken.

12.4.5 *Reproducibility*—The pooled reproducibility of the STSA Method has been established as $2.59 \times 10^3 \text{ m}^2/\text{kg}$ ($2.59 \text{ m}^2/\text{g}$). Two single test results (or determinations) produced in separate laboratories that differ by more than $2.59 \times 10^3 \text{ m}^2/\text{kg}$ ($2.59 \text{ m}^2/\text{g}$) must be considered suspect, that is, that they represent different sample populations. Such a decision dictates that appropriate investigative or technical or commercial, or both, actions be taken.

TABLE 2 NSA and STSA Values of SRB-5 Materials

Standard	NSA, m ² /g	STSA, m ² /g
SRB A5	141.5	124.3
SRB B5	74.3	73.6
SRB C5	122.0	111.7
SRB D5	27.5	27.1
SRB E5	35.3	34.2
SRB F5	39.1	37.1
SRB G5	9.1	8.4

TABLE 3 ASTM Test Method Precision—Type 1 Carbon Black: NSA^A

Material	Mean Level 10 ³ m ² /kg (m ² /g)	Within Laboratories ^B			Between Laboratories ^B		
		Sr	r	(r)	SR	R	(R)
SRB D-4 (N762)	24.1	0.21	0.61	2.52	0.47	1.34	5.57
SRB E-4 (N660)	34.6	0.24	0.67	1.92	0.40	1.13	3.28
SRB F-4 (N683)	38.5	0.18	0.51	1.32	0.62	1.74	4.53
SRB B-4 (N330)	74.4	0.43	1.22	1.64	0.90	2.55	3.43
SRB A-4 (N326)	77.3	0.47	1.32	1.71	0.93	2.64	3.42
SRB C-4 (N121)	124.2	0.73	2.06	1.66	1.76	4.99	4.02
Pooled or average values	62.2	0.42	1.19	1.92	0.96	2.72	4.38

^AThis is short-term precision (days) with: p = 7, q = 6, and n = 4.

^BSymbols are defined as follows:

- Sr = within laboratory standard deviation,
- r = repeatability (in measured units),
- (r) = repeatability (in percent),
- SR = between laboratory standard deviation,
- R = reproducibility (in measured units), and
- (R) = reproducibility (in percent).

TABLE 4 ASTM Test Method Precision—Type 1 Carbon Black: STSA^A

Material	Mean Level 10 ³ m ² /kg (m ² /g)	Within Laboratories ^B			Between Laboratories ^B		
		Sr	r	(r)	SR	R	(R)
SRB D-5 (N762)	27.1	0.56	1.58	5.84	0.72	2.04	7.51
SRB E-5 (N660)	34.2	0.52	1.47	4.30	0.71	2.01	5.87
SRB F-5 (N683)	37.1	0.66	1.87	5.03	0.78	2.21	5.95
SRB B-5 (N330)	73.6	0.67	1.90	2.57	0.74	2.09	2.84
SRB C-5 (N220)	111.7	0.91	2.57	2.30	1.24	3.51	3.14
SRB A-5 (N135)	124.3	0.94	2.66	2.14	1.15	3.25	2.62
Pooled or average values	68.0	0.73	2.06	3.03	0.92	2.59	3.81

^AThis is short-term precision (days) with: p = 7, q = 6, and n = 4.

^BSymbols are defined as follows:

- Sr = within laboratory standard deviation,
- r = repeatability (in measured units),
- (r) = repeatability (in percent),
- SR = between laboratory standard deviation,
- R = reproducibility (in measured units), and
- (R) = reproducibility (in percent).

Report repeatability and reproducibility in percent.

12.5 *Bias*—In test method terminology, bias is the difference between an average test value and the reference (true) test property value. Reference values do not exist for this test method since the value or level of the test property is exclusively defined by the test method. Bias, therefore, cannot be determined.

NOTE 3—The nitrogen surface area of dry process (pelletized with oil) carbon blacks is typically lower than wet process (pelletized with water) and un-pelletized carbon blacks. The oil used in the manufacture of dry process carbon blacks typically suppresses the nitrogen surface area measurement.

13. Keywords

13.1 carbon black, B.E.T.; Brunauer/Emmett/Teller; nitrogen adsorption; nitrogen surface area; surface area by multi-point B.E.T. method, external surface area; statistical thickness surface area

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